## **General chemistry**

(შუალედური გამოცდის ნიმუში)

	Ouestion $\bf 1$ What is the relative molecular mass of ZnO and $H_2SO_4$ ?
0000	Select one: a. 58 and 87 b. 37 and 28 c. 81 and 98 d. 64 and 98
0000	Ouestion 2 What is the mass of 2.5 mole Fe? Select one: a. 40 b. 155 c. 100 d. 140
0000	Ouestion 3 How many molecules are there in a 5g sample of sulphuric acid ( $H_2SO_4$ )? Select one: a. $0.3 \times 10^{23}$ b. $5. 2 \times 10^{23}$ c. $6.2 \times 10^{22}$ d. $0.2 \times 10^{24}$
0 0 0	Ouestion 4  How many protons and neutrons are in atom of 31Ga?  Select one:  a. 31 and 29;  b. 31 and 39;  c. 31 and 45  d. 31 and 70;
0	Give the group number and period number for elements 20Ca, 80Hg.  Select one:  a. 4 period. Ia group, 5period IIb group;

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O	b. 4 period. Ha group, 6 period Hb group;
0	c. 3 period. Ia group, 4period IIa group;
0	d. 2 period. Ia group, 5period IVb group;
	Question 6 Which of the following compounds are formed by covalent bond?
	Select one:
0	a. H <sub>2</sub> O NaCl Cl <sub>2</sub>
0	b. N <sub>2</sub> HCl H <sub>2</sub>
0	c. KCl HBr CCl₄
0	d. H <sub>2</sub> S AICI <sub>3</sub> LiCI
	Question 7 Wich atoms are connected by non-polar covalent bond?
	Select one:
0	a. Br, Br;
0	b. , Cl
0	c. H, CI;
0	d. Mg, O;
	Question $\bf 8$ What type of hybridization is required at the central atom of the following molecules: $CH_4$ ,
0	What type of hybridization is required at the central atom of the following molecules: CH <sub>4</sub> ,
0	What type of hybridization is required at the central atom of the following molecules: CH <sub>4</sub> , Select one:
000	What type of hybridization is required at the central atom of the following molecules: CH <sub>4</sub> , Select one:  a. sp
0000	What type of hybridization is required at the central atom of the following molecules: CH <sub>4</sub> , Select one:  a. sp  b. sp <sup>2</sup>
0 0 0	What type of hybridization is required at the central atom of the following molecules: CH <sub>4</sub> , Select one:  a. sp b. sp <sup>2</sup> c. sp <sup>3</sup>
0000	What type of hybridization is required at the central atom of the following molecules: CH <sub>4</sub> , Select one:  a. sp b. sp <sup>2</sup> c. sp <sup>3</sup> d. sp <sup>3</sup> d
000	What type of hybridization is required at the central atom of the following molecules: CH <sub>4</sub> , Select one:  a. sp b. sp <sup>2</sup> c. sp <sup>3</sup> d. sp <sup>3</sup> d  Question 9
0 0 0	What type of hybridization is required at the central atom of the following molecules: CH <sub>4</sub> , Select one:  a. sp b. sp <sup>2</sup> c. sp <sup>3</sup> d. sp <sup>3</sup> d  Ouestion 9  which row contains only acidic oxides:
000	What type of hybridization is required at the central atom of the following molecules: CH <sub>4</sub> , Select one: a. sp b. sp <sup>2</sup> c. sp <sup>3</sup> d. sp <sup>3</sup> d  Ouestion 9  which row contains only acidic oxides: Select one:
000	What type of hybridization is required at the central atom of the following molecules: CH <sub>4</sub> , Select one:  a. sp b. sp <sup>2</sup> c. sp <sup>3</sup> d. sp <sup>3</sup> d  Ouestion 9  which row contains only acidic oxides: Select one:  a. K <sub>2</sub> O,SiO,SO <sub>3</sub>
0 0 0	What type of hybridization is required at the central atom of the following molecules: CH <sub>4</sub> , Select one:  a. sp b. sp <sup>2</sup> c. sp <sup>3</sup> d. sp <sup>3</sup> d  Cuestion 9  which row contains only acidic oxides: Select one:  a. K <sub>2</sub> O,SiO,SO <sub>3</sub> b. CuO,Na <sub>2</sub> O,NO
000	What type of hybridization is required at the central atom of the following molecules: CH <sub>4</sub> , Select one:  a. sp b. sp <sup>2</sup> c. sp <sup>3</sup> d. sp <sup>3</sup> d  cuestion 9  which row contains only acidic oxides: Select one:  a. K <sub>2</sub> O,SiO,SO <sub>3</sub> b. CuO,Na <sub>2</sub> O,NO c. CO,N <sub>2</sub> O,NO d. N <sub>2</sub> O <sub>3</sub> ,P <sub>2</sub> O <sub>5</sub> ,CO <sub>2</sub> cuestion 10
000	What type of hybridization is required at the central atom of the following molecules: CH <sub>4</sub> , Select one:  a. sp b. sp <sup>2</sup> c. sp <sup>3</sup> d. sp <sup>3</sup> d  Cuestion 9  which row contains only acidic oxides: Select one:  a. K <sub>2</sub> O,SiO,SO <sub>3</sub> b. CuO,Na <sub>2</sub> O,NO c. CO,N <sub>2</sub> O,NO d. N <sub>2</sub> O <sub>3</sub> ,P <sub>2</sub> O <sub>5</sub> ,CO <sub>2</sub>

0000	a. HCI,NH <sub>3</sub> ,H <sub>2</sub> CO <sub>3</sub> b. Cu(OH)CI,H <sub>2</sub> SO <sub>2</sub> ,H <sub>2</sub> SO <sub>3</sub> c. HCI,AI(OH) <sub>2</sub> CI,NH <sub>2</sub> OH d. KCI,NaHSO <sub>4</sub> ,HNO <sub>3</sub>
	Question 11 which row contains only sulfides :
0000	Select one:  a. Na <sub>2</sub> S,CuS,FeS  b. K <sub>2</sub> SO <sub>3</sub> ,NaHSO <sub>3</sub> ,Al <sub>2</sub> (SO <sub>3</sub> ) <sub>3</sub> c. ZnSO <sub>4</sub> ,KHSO <sub>4</sub> ,NaSO <sub>4</sub> d. K <sub>2</sub> S <sub>2</sub> O <sub>3</sub> ,Na <sub>2</sub> S <sub>2</sub> O <sub>3</sub> ,CaSO <sub>3</sub>
	Cuestion 12 The coordinate number of the central atom in the complex compound $K_2[PtCl_4]$ is: Select one:
0000	a. 4 b. 6; c. +3; d. 3;