## General chemistry <br> 

Question 1
What is the relative molecular mass of ZnO and $\mathrm{H}_{2} \mathrm{SO}_{4}$ ?
Select one:
a. 58 and 87
b. 37 and 28
c. 81 and 98
d. 64 and 98

## Question 2

What is the mass of 2.5 mole Fe ?
Select one:
O a. 40
b. 155
c. 100
d. 140

Question 3
How many molecules are there in a 5 g sample of sulphuric acid $\left(\mathrm{H}_{2} \mathrm{SO}_{4}\right)$ ?
Select one:
a. $0.3 \times 10^{23}$
b. $5.2 \times 10^{23}$
c. $6.2 \times 10^{22}$
d. $0.2 \times 10^{24}$

## Question 4

How many protons and neutrons are in atom of 31 Ga ?
Select one:
a. 31 and 29;
$\bigcirc$
b. 31 and 39;
c. 31 and 45

0
d. 31 and 70;

Question 5
Give the group number and period number for elements ${ }_{20} \mathrm{Ca},{ }_{80} \mathrm{Hg}$.
Select one:
C a. 4 period. Ia group, 5 period IIb group;

O
b. 4 period. IIa group, 6 period IIb group;
c. 3 period. Ia group, 4period IIa group;

O
d. 2 period. Ia group, 5 period IVb group;

Question 6
Which of the following compounds are formed by covalent bond ?
Select one:
a. $\mathrm{H}_{2} \mathrm{O} \mathrm{NaCl} \mathrm{Cl} 2$
b. $\mathrm{N}_{2} \mathrm{HCl} \mathrm{H}_{2}$
c. KCl HBr CCl 4

C
d. $\mathrm{H}_{2} \mathrm{~S} \mathrm{AlCl}_{3} \mathrm{LiCl}$

Question 7
Wich atoms are connected by non-polar covalent bond?
Select one:
a. $\mathrm{Br}, \mathrm{Br}$;
b. , Cl

0
c. $\mathrm{H}, \mathrm{Cl}$;

C
d. Mg, O;

Question 8
What type of hybridization is required at the central atom of the following molecules: $\mathrm{CH}_{4}$, Select one:
a. sp
b. $\mathrm{sp}^{2}$

O
C. $\mathrm{sp}^{3}$

0
d. $s p^{3} d$

Question 9
which row contains only acidic oxides:
Select one:
a. $\mathrm{K} 2 \mathrm{O}, \mathrm{SiO}, \mathrm{SO}_{3}$
b. $\mathrm{CuO}, \mathrm{Na}_{2} \mathrm{O}, \mathrm{NO}$
c. $\mathrm{CO}, \mathrm{N}_{2} \mathrm{O}, \mathrm{NO}$
d. $\mathrm{N}_{2} \mathrm{O}_{3}, \mathrm{P}_{2} \mathrm{O}_{5}, \mathrm{CO}_{2}$

Question 10
which row contains only acids:
Select one:

C a. $\mathrm{HCl}, \mathrm{NH}_{3}, \mathrm{H}_{2} \mathrm{CO}_{3}$
b. $\mathrm{Cu}(\mathrm{OH}) \mathrm{Cl}, \mathrm{H}_{2} \mathrm{SO}_{2}, \mathrm{H}_{2} \mathrm{SO}_{3}$
c. $\mathrm{HCl}, \mathrm{Al}(\mathrm{OH})_{2} \mathrm{Cl}, \mathrm{NH}_{2} \mathrm{OH}$

C d. $\mathrm{KCl}, \mathrm{NaHSO}_{4}, \mathrm{HNO}_{3}$
Question 11
which row contains only sulfides:
Select one:
O. $\mathrm{Na}_{2} \mathrm{~S}, \mathrm{CuS}, \mathrm{FeS}$
b. $\mathrm{K}_{2} \mathrm{SO}_{3}, \mathrm{NaHSO}_{3}, \mathrm{Al}_{2}\left(\mathrm{SO}_{3}\right)_{3}$
C. c. $\mathrm{ZnSO}_{4}, \mathrm{KHSO}_{4}, \mathrm{NaSO}_{4}$

O
d. $\mathrm{K}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}, \mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}, \mathrm{CaSO}_{3}$

Question 12
The coordinate number of the central atom in the complex compound $\mathrm{K}_{2}\left[\mathrm{PtCl}_{4}\right]$ is:
Select one:
a. 4
b. 6;
$\bigcirc$
c. +3 ;

0
d. 3;

